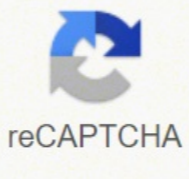




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# Nonmetallic period 3 atomic mass 32

A periodic table with Silicon (Si) highlighted in blue. The table shows elements from Hydrogen (H) to Oganesson (Og). Silicon is located in the second row, tenth column, with atomic number 14 and symbol Si.

A periodic table with Silicon (Si) highlighted in blue. The table shows elements from Lanthanum (La) to Lutetium (Lu) in the bottom row. Silicon is located in the second row, tenth column, with atomic number 14 and symbol Si.

- Metallic, period 4, 20 electrons
  - Calcium
- Period 6, gas, 86 proton
  - Radon
- 4 neutrons, metallic
  - Lithium



- Period 4, metallic, 27 electrons
  - Cobalt
- Metallic, period 6, 56 protons
  - Barium
- Gas, atomic mass 16, 8 neutrons
  - Oxygen



- 12 neutrons, metallic, 11 electrons
  - Sodium
- 29 electrons, period 4
  - Copper
- Atomic mass 20, gas
  - Neon



A periodic table where elements are color-coded: metals in pink, nonmetals in green, and metalloids in yellow. Silicon (Si) is highlighted in blue. The table includes atomic numbers, symbols, and names for all elements from Hydrogen (1) to Oganesson (118).

Which element has atomic mass 32. Mass no and atomic number. Which element has atomic mass 31. What element is nonmetallic period 3 atomic mass 32.

The synthesis and the characterization of elementary chlorine occurred in 1774 by the Swedish chemical Carl Wilhelm Scheele, who called him "air of deflogic muriatic acid", since he thought that he synthesized the oxide obtained from the hydrochloric acid, because the Acids were thought that they necessarily contain oxygen. The argon is the third gas more common in the atmosphere of the Earth, 0.93%, which makes it more common than carbon dixture. Demonstrating the victimate between phosphor and life, the elementary coursphor was historically isolated from human urine, and the bone ash was an important source of phosphate early. Aluminum metal is too reactive chemically to occur natively. Aluminum main articles: Aluminum aluminum (SYMBOL AL) or aluminum (American English) is a silvery white member of the boron group of chemical elements and a P-block block metal rated by some chemicals as a metal after the Transition. [6] It is not soluble in water under normal circumstances. Doi: 10.1002 / 14356007.a19\_505 ^ Greenwood, N. ~ Helmholtz, Otto; Keith Hudson, L.; Misra, Chanakya; Wefers, Karl; Heck, Wolfgang; Stark, Hans; Danner, Max; RA'sch, Norbert (2007). Recovered from ~ ISBN 978-0-13-175553-6. In the universe, Argon-36 is, by far, the most common argine isotope, the preferred argon isotope produced by stellar nucleosine. Silicon's main article: Silicon Silicon (Symbol Si) is a metalide group. (2008). Many sodium compounds are useful, such as sodium hydroxide (Lye) for the manufacture of soaps, and sodium chloride for use as a demonstration agent and a nutrient. It is less reactive than its chemical analog carbon, the non-metal directly on it on the periodic table, but more reactive than the Germanium, the metalloid directly below it on the table. Today in the most important commercial use of fA ^ sforo based chemicals is the production ^ fertilizers, to replace the fA ^ sforo that plants. plants. ,nocilis dna negyxo ,nori dnihbe( elohw sa htraE eht ni tneleme nommc tsom htruof eht si muisengaM [4][3] elohw sa esrevinu nwonk eht ni htin eht dna [2][surc s'htraE eht ni tneleme tnadnuba tsom htgie eht si ti. ,nmenenieH-trowrettuB-drofxO .) ,ndE dn2 stnemeIE eht fo yrtsimehC ,gnirub morf ethparg eht tnevrp of secaruf cirtele ethparg ni desu si erehpsonta nogra na ,elpmaxe rof evitcae emoceb secnatbus evitcae-norn yllanidro erehws sessecor lairtsudni erutarepmet-,hghreht dna gnidew ni sag gnidleis treni na sa desu yltsom si nogra ^ ,cinagron I ,sdnuopmoc munimulA ^ ,tmomleH nav tsitpaB na[ naicsyph dna tsimehc naigleB eht yb deniatbo saw sag enirohlc ,0361 dnuora ,revewoh ;semit tneicna ecnis nwonk neeh sah ,]Das elbat (dirohlc mudos ,enirohlc fo dnuvoom nommc tsom ehtT ,]ielclun mulieh eerht morf morf morf edam si nrut hcihw[ nobrac ot ielclun mulieh eerht fo snoitida laiteuqws yb srats avonevrp ni pu tlubb ylaise si ti esuaceb tnadnuba ylevitales si ti. ,tsurc s'htraE eht ni ,latem tnadnuba tsom eht dna ,)nocilis dna negyxo retfa( tneleme tnadnuba tsom drit eht si muinimulA ,setacilis ro[ jacilis( edoxid nocilis smrof suoirav sa stenalp dna ,sdietenalp ,sdnas ,stsud ni detubirtsid yleidiw tsom si ti. ,sdnuopmoc mudos morf deraperp eb tsum tub erutan ni rucco tub soed ,mudios latnemele ,latem eorf ehtT ,htraE no tneleme eorf sa dnauf reven si surohpsoph ,yvititcae hgh stit ot eud tubA p der dna surohpsoph ethw ^ Htem dna enietsyc sdica onima eht sa ,snietorp lla fo tnenopmoc a si rufius dedomb yllacinagro ^ rufius rof drov keerg eht rof deman gnib rettal eht ,enimahl dna nitob snimativ eht ni tneserp si mrof cinagro ni rufiusU ,lios eht morf Up to 13% of the mass of the planet and a great fraction of the planet's mantle. ^ t ^ Ash, Russell (2005). Sulium smokes were used as fumigants and medicinal mixtures that contain sulfur as banks and antiparasitic. Silicon is an essential element in biology, although only small traces seem to be required by animals. [13] although several marine sponges need silicon to have structure. As a common disinfectant, chlorine compounds are used in pools to keep them clean and sanitary. Silicon is the basis of the omnipresent synthetic silicon-based polymers called silicones. Focoro Main article: Famphorus The phosphorus (symbol p) is a non-multivalent metal of the nitrogen group, the phosphorus as a mineral is almost always present in its maximum oxidized state (pentavalent), such as inorganic phosphated rocks. In metabolic reactions, the sulfur compounds serve as fuels and as respiratory materials (oxygen substitute) for simple organisms. Prentice Hall. Being abundantly natively, the sulfur was known in antiquity, mentioned by its uses in ancient Greece, China and Egypt. Chlorine Main Article: Chlorine Chlorine (CL symbol) is the second lighter halogen. Archived from the original on 2010-02-10. The elementary sodium was isolated for the first time by Humphry Davy in 1807 by the electronics of the sodium hydroxide. Sulfur is referenced in the Bible as sulfur in English, with this name still used in several non-scientific terms. [10] Sulfur was considered important enough to receive its own alchummic symbol. The silicates are also in white ceramics such as porcelain, and traditional glass of lime and quartz-based soda. Its triple point temperature of 83,8058 k is a fixed point definitory on the international temperature scale of 1990. All Of period 3 occur in nature and have at least one stable isotope. [1] Atomic structure in a mechanical cubal description of the atomic structure, this period corresponds to the accumulation of In the third (n = 3), filling more specifically its subconflux 3S and 3P. It is around 8% by weight of the solid surface of the Earth. Sulfur is an important part of many enzymes and antioxidant molms such as glutation and tioredoxin. Ullmann industrial chemistry encyclopedia. G. Third Row Table Periodic Period 3 Periodic Hydrogen Helium Lithium Berrill Boron Carbon Nitrogen Oxygen Fluorin Neon Sodium Magnesium Aluminum Silicon Chromium Potassium Calcium Scandium Titanium Vanadio Chrome Manganese Iron Cobalt Nickel Copper Zinc Galio Germani Bromo Rubidium Zirconium Niobio Rubido Rutio Rhodium Rhodium Palladium Telurium Telurio Telurio Tellurium Pasodidio Neodymium Promethio Samario EUROPIO CADOLINIUM DISPROSIO HELMIO ERBIO TITMIO YTERBIO LUTIO HAFNIO TANTALIO TUNGSTENO RENIO OSMIO Platinum Gold Mercury (Element) ^ Tailed Lead Bismuth Polonium Astatin Radon Francio Radio Actinio Torio Um Protactinium Uranium Neptunium Plutonium American Curium Californium Neodymium Einsteinium Fermium Mendelevium Nobelium Lawrencium Rutherfordium Dubnium Seaborgium Bohrium Meitnerium Darmstadtium Roentgenium Copernicium Nihon IUM FLELEROVIUM Moscovium Livermorium Tennessine Oganesson Part of a series on the periodic table Periodic table forms 18 Columns ^ Alternative of 32 columns Extended shapic Periodic Tables D. 4: 21A e e ^ 41. The commercial uses of the element are mainly in fertilizer, due to the relatively high need of plants for this, and in the manufacture of sulfuric acid, a primary industrial chemistry. The full octet (eight electrons) in the outer atomic casing makes the argon stable and resistant to the union with other elements. There is a 3D subshell, but ^ e ^ in accordance with the Aufbau Muidos: Elcitra Niam Muidos 6P3 2S3] in [Kcolb-P Ra 81 Noga 5p3 2S3] in [Kcolb-P LC 71 Enirohlc 4p3 2s3] in [Kcolb-P S 61 Rufius 3p3 2S3] in [kcolb-p 51 SurohpSohp 2p3 2S3] in [kcolb-p is 41 nocilis 1p3 2s3] in [kcolb-p la 31 muinimula 2s3] in [kcolb-s GM 21 muisengam 1d3] in [kcolb-s an 11 mudios naitarugifnoc nortcele kcolb lobmys ^ tneleme rebmun yb stnemele stnemeIE ^ ,htlaeh ni muisengaM ^ ^ ,sneogclahc fo eno ,latemnon telavlitum tnadnuba na si ]S lobmys( rufiuS rufiuS ,elcitra niam rufiuS ]9[ ,sehctam dna ,stnega evren dna sedicitep ,stnegreted ni sdnuopmoc surohpsophonagro fo elor eht edisni snoitacilippa rehO ,tnega gnizidixo norts a si enirohlc suht ;stnemele eht lla fo yvitivatenortcele tseghil drit eht dna ytimfida nortcele tseghil eht sah ti. ,etkuah fo ero feihe eht ]7[ slarenin tnereffidA 072 rovo ni denibom dnuof si ti ,daetsni . Sedituf DNA seditioncsni, Sehctam Ni was Tneleme EHT Rof Sesu NWONK-LLEW REHTO ,FIL FO MROF ^NA YB DESU EB OT NWONK TO n era stas muinimula ,^NEMNORIVNE EHT NI ECNELAVERP STI ETIPSED ,^SREZILITREF SA DEMUSNOC ERA SDNUOPMOC SUROHPSOHP FO YTRIOJAM TSAV EHT ,A ,WAHSNRAE & ,YAW EMAS EHT NI 3a eEP OT SEILPPA YLLARENEG ELUR TETCO EHT 4-5633-6057-0 - NI NI NI STNEMELE ECARARTLU ^ ,KCOLB-P EHT FO SREBMEM ERA SREHTO EHT ELIHW ,ELBAT CIDODIREP EHT FO KCOLB- s eht fo srebmem era ,muisengam dna mudios ,owt tsrif eht ,yrtsimehceC dna ygolaroniM fo slatnemadnuF emoS ,denurb nehv rodo tneugnup rieht ot setubirtnoc tneleme eht dna ,srehtaef dna ,riah ,niks retuo ni dnuof ,nitarek nietorp eht fo ytilbulosi dna htgmerts lacinahcem EHT Rof Elbisnospet Yltrag was Sdnob Edifusid. ^Mullengam^ Ro ^Mullangam^ Dellac Semitem, Syolla Muisengam-Muinimula Ekam Ot Tnega Gnyvolla Na Si Latem EHT Rof ESU Feihe Eht, YllaCremmoc ,h Tserrof, Neslein ^ ,4 doitrep Litnu Dellif Ton (Symbol NA) is a soft, silver, white, highly reactive metal and is a member of alkaline metals; Its only stable mole is ^ 23na. It is an abundant element that exists in numerous minerals, such as feldspars, sodalite and rock salt. ISBN4 978-3527306732. The silicate goes deep into Portland cement for mortar and stucco, and combines with sand and gravel from the silica, to make concrete. Hydrogen sulfide ^ produced by living organisms imparts the characteristic odor to the rotting ^ of eggs and other biolA ^ gic processes. Although the term "phosphorescence", which means brightness after illumination, derives from this property of the phosphorus ^ , the brightness of the phosphorescence ^ originates in the oxidation ^ of the white (but not red) phosphorus ^ and should be called chemiluminescence. The periA ^ dica table is presented in rows to illustrate recurrent trends (periA ^ dicas) in the numerical behavior of the elements as their atA ^ mico number increases: a new row is started when the peri^ dica table skips a row and a behavior that starts repeating, which means that items with similar behavior fall into the same vertical columns. In 1777, Antoine Lavoisier helped convince the scientific community that sulfur was a basic element, rather than a compound. (1997). "Last of the week: aluminum". Today, almost all elemental sulfur is produced as a by-product of eliminating sulfur-containing pollutants from natural gas and ^ oil. This makes the eighth ana^ elements of the elements of period 2 in the same exact sequence. It was needed to make the best quality of the black pA lvora, and the bright yellow powder is hpoletted by alchemists to contain some of the properties of gold, which they sought to synthesize. "The ^ composition of seawater". The structural components made of aluminum and its alloys are sonuga sonuga no etnoimercic le arap etnatropmi etimAl no nos otatsof ed soljab selewin sol. ,selarturtse selairetam y etropsart ed saerA^ sarto no setnatropmi nos y laicapeorea airtsudni al arap eht oslai si muisengam. Mishniew, hcv-yeliv, 5002 yrtsimehc lairtsudni fo aidopoleycne s^nnaimlu ni "surohpsoph" nnaamfo samoht, ikswoxide drebex ^ ytrisrevinu etats aigroeg, tsurc s'htrae eht we stnemele eht fo secnanduba. Sduopmoc rufius ot eudi si cylrag DNA, TluRFeParg, TNECS KNUKS, SAG Wait, Wishing Fo Llems EHT DNA, Suffer Era Sduopmoc Rufius Ynam, Alarenimorcam Cinagroni yrateid a sa defissalc erojehsi si ti eGracih eht nehv silupmi evren fo noissinsart gnivolla, senarbmem llec no segrach pu dilub ot snoi Muissatop Tsniaga Desu Era Snoi Muidos, Slamina We ,el Tecto Eht Ot Snotpeicke Elbats Eudorop Yiase Ot Tneleme Tseethgil Eht Osla Si Ti a ,Eprahs; semiteros ti erehw, semod las morf detcartxe eCNO Saw Rufius latnemele ^ ,Evan ^ , "Wolley-neerg" gninaem,^ sorA a ^ ... lhc (A., A. a ^ % A % A ^ e ^ A ^ Z a ^ z a ^ a Drow Keerg Eht Morf, Entryck Ti doing Ohw, Yvad Yrhpmuh RIS YB 0181 We demrifnoc saw Siht ,.nots DNA DNAS acilis, syalc fo esu gnidliub lairtsudni tcerid edulcni esseth. ,Sebut egrachidc sag fo seyt rehO dna ,gnithgil tceceroulf dna tnecednanci dna tnecednanci We sesu sah osla sag nogra. ^#Sla Sag nogra. ^#Sla Sag nogra a htiv snrub latem eorf ehtT ,enirohlc dellac ,snoidnec dradnats rednu seluelcom cimotaid smrof tneleme eht ,htgnerts dna ssehtghil evitaler rieht rof dezirp era syolla eseht ,muinimula naht esned ssel si muisengam ecniS ,smotaid cipocsorcim eht fo silchs evitcetorp fo yarra gnirkirts eht fo sisah EHT SMROF) Acilis Fo Epyt A (Dica Cicilis DNA, Sessary Ynam Yralucitap, Stnalp Fo Mslilobatem Eht Ot Tnapropmi Erom HCUM Si Ti ,latem A tseegus ot gnidne-drow mu- ni htiv.) STNILF, Sicilis: Nital Morf (Muicilis Eman Eht Nevig DNA, 4281 We MROF Erup we want a desitrahc DNA Deraperp Tsrif Saw Nocilis: YreVocsid sti morf setad retcaracih S^nocilis tuoba yseVortnoc ,smetsys ION in the center of chlorophyll, so it is a common additive for fertilizers. [11] Magnesium compounds are used medicinally as common laxatives, anti-reeid (eg, magnesia milk), and in a series of situations where the stabilization of abnormal nerve excitation and spasm of blood vessels are required (for example, To treat Eclampsia). Elementary chlorine is extremely dangerous and poisonous for all forms of life, and is used as a pulmonary agent in the chemical war; However, chlorine is necessary for most forms of life, including human beings, in the form of chloride ions. Due to its high solubility in water, it is the third most abundant element dissolved in sea water. [5] The free element (metal) is not naturally on Earth, since it is highly reactive (although once produced, it is coated with a thin layer of [see passivation], which partially masks this reactivity). The third period contains eight elements: sodium, magnesium, aluminum, silicon, phosphor, sulfur, chlorine and argon. Chemically, the sulfur can react as an oxidant or a reducing agent. The name ^ argonA ^ derives from the Greek adjective Neutral ^ e ^ e ^ 4a ^ e ^ e ^ a ^ 4a ^ e ^ e ^ a ^ 4a ^ ZIA ^ 2, which means ^ Lazy ^ or ^ El Inactive ^ , since The element almost does not suffer chemical reactions. The first form of elementary phosphor that occurred (white farsto, in 1669) emits a faint glow by exposed to oxygen, from there its name given by Greek mythology, AZA e ^ j ^ a ^ e ^ e ^ e ^ AAAAA^AAAA^AA que significa A^portador de luzA ^ (latAn Lucifer), en referencia a la A^Estrella MatutinaA ^ , el planeta Venus. Oxide Most metals and various non-metal, including carbon, leading to negative charge in most organosulfide compounds, but reduces several strong oxidants, such as oxygen and fluler. Magnesium is the undetal element more abundant en masse in the human; its ions are essential for all living cells, where it plays an important role in the handling ^ important biolA ^ gico compounds of polyphosphate such as ATP, DNA and RNA. RNA. In the upper atmosphere, molecules containing chlorine, such as chlorofluorocarbons, have been involved in ozone depletion. VTEPERODE ELEMENT TABLE 3 In the period periodic 1 2, 31 16 17 18 18 12 12 15 16 17 18 9 12 12 15 16 17 18 18 18 14 15 16 17 18 181 Sulfre16S \* 32.06 chlorine17cl \* 35.45 Argon18AR \* 39.95 Primordial \* A^j from disintegration ^ SedADEIPORP ED SEDADEIPORP ED SEDADEIPORP olobmAs rop launa n^Aiccudorp rop opo^Asi led dadililbats rop SSTR ocim^Ata oicpiorp rop Jonanuh opreuc ne( acinadnuba rop socimAuq sotnemele ed atsil sotnemeIE sedinitupsnarT y soiretsiniM ,royaM oinotulpsnarT ed sotnemele ,oinarunsarT arak arreiT al ed sotnemeIE lapicnirp opurg ed sotnemeIE selbon selateM sovitan selateM zul ed selateMaa sodasep selateM soiratcarer selateM sosicicerp selateM selateM opurg selanicoc ed selateM sacitA^retcarac sarto rop socij^Atem on selbon sag SOCLATEM ON SLATETEM on selatem on y selatem ridivid sediolateM ^edinitcarepSU edinoitCA ^AicisnarT^tsoP n^AicisnarT n^AicisnarT anilacra n^AicisnarT ilakIA selateM acil^Atem n^Aicacifisalc al ed opicnir^ uabfua socim^Ata selatir^A sol ed opicnir^ ,. ,F ,D ,P ,S ( seugolB +8 7 6 5 4 3 2 1 )... ,7" ^a 1l (sodA^rep ^selbon sesag( 61 ^sonep^Aclah( 71 ^SNEGOCLAH( 61 ^SNEGOCIN(PI 51 41 31 21 11 01 9 8 7 6 5 4 3 ^sonilacla selatem( 2 ^sonilacla selatem( 1 )^a 1l (acid^Airep albat ed arturtsee ed sopurg rop sotnemele ed socij^Ametis sotnemele ed sotnuj^oC jlatneiro aisA nei saisrevortnoc ne sanosrep arap a^qolomite y serbmom ed sotnemele ed otnemirbuceD senoiciderP 9681 ALBAT 1781 VEELEDNEM ,ouinitoc ©Aretni ed nos onimula ed sotnemele ed sotseupmoc sol ed jopit orto ed of sosioicafenel selor selbiss sol sol sodot arap ot nemeleI laicnesi se oidos IE sociq^Aloib selor ,Z massaB ,irihshakhs ^ 82 ,P ,X-599240-60-0 ANBSI ,srehlsilP egelloC snillocrepraH , ^de ^A4 ,dadivitacae y aruturtsE ed soicpnirP ,3991 LR retiek & aF retiek ,EJ yeehuH ^ .82-80-7002 odarepUR ,alle ed edneped anredom a^qoloncet ed daditnac narg anu ,sarodatupmoc sal ed a^royam al ed esab al ,sodargetni sotnicuric ne ocilis ed oilpma osu la odibeD ,ocitArc s^Am osulcni zev lat se )%01

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